Triterpenoids from the Stem Bark of *Avicennia officinalis*

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**ABSTRACT:** The triterpenoids, betulinic acid, lupeol and betulinaldehyde, were isolated from the ethyl acetate extract of the stem bark of *Avicennia officinalis* (Avicenniaceae) by a combination of column and preparative thin-layer chromatography over silica gel. The structures of these compounds were determined by spectroscopic analysis (UV, IR, $^1$H NMR, $^{13}$CNMR and EIMS). This is the first report of a systematic phytochemical investigation and the presence of these triterpenoids from this plant.

**Key words:** Triterpenoid, Avicenniaceae, Betulinic acid, Lupeol and Betulinaldehyde

**INTRODUCTION**

*Avicennia officinalis* is a medium-sized tree growing in brackish water. The 15 species in the single genus of Avicenniaceae family are found on tropical coasts as constituents of mangrove vegetation. Previous Phytochemical investigations on the different species of *Avicennia* resulted in the isolation of essential oil and sugars like arabinose, glucose and ribose. Among other compounds alkaloids, flavonoids, steroids, terpenoids and iridoids are most considerable components. In Bangladesh, *Avicennia officinalis* is widely distributed in Sundarban and locally it is known as Baen. This plant is used for thrush in children. The heartwood is rubbed against a course stone. The tree oils of this plant exhibited cytotoxic activity. The earlier studies on this plants resulted in the isolation of C iridoid glucoside, 7-O-trans cinnamoyl-4-epilogenin, geniposidic acid, 2-cinnamoylmussaenoside. So far no detail phytochemical and biological studies have been carried out on this plant. Since this plant has good medicinal properties, the present work has been undertaken to isolate, purify and identify secondary metabolites. In this paper the isolation and structural elucidation of the betulinic acid (1), lupeol (2) betulinaldehyde (3) by using spectroscopic techniques like UV, IR, $^1$H NMR, $^{13}$CNMR and EIMS are being reported.

**MATERIALS AND METHODS**

**General.** Melting points were determined on a kolfer hot-stage apparatus and are uncorrected. UV spectrum was taken in MeOH solution using a Perkin-Elmer lambda 9UV/Vis./NIR Spectrometer. IR spectra were recorded on CHCl$_3$ solutions on either a Perkin-Elmer 580 or Philips 9800 FTIR Spectrometer. $^1$H NMR and $^{13}$C NMR spectra were obtained on Bruker WP 200 SY and AM 200 SY instruments ($^1$H, 200. 132 MHz; $^{13}$C, 50.32 MHz) using TMS as internal standard and CDCl$_3$ as solvent. Electron impact mass spectra (EIMS) were recorded using a VG updated MS 12 Spectrometer and optical rotations were measured on an optical activity AA-
100 Polarimeter in CHCl₃ solutions at 20°C. Petroleum ether Specifically refers to the bp 40-60°C fractions.

**Plant materials.** The stem bark of *Avicennia officinalis* Gaertn was collected from Khulna district of Bangladesh. A voucher specimen has been deposited at the Herbarium of the University of Glasgow, Glasgow, U.K.

**Extraction and isolation.** The Sun-dried stem bark powder (500 g) of *A. officinalis* was extracted in a Soxhlet apparatus for three days with EtOAc. This extract was concentrated in vacuo and subjected to flash column chromatography over silica gel (Merck Kieselgel GF₂₅₄). Elution of the column first with petroleum ether, increasing amounts of EtOAc in petroleum ether and finally with methanol yielded a number of fractions. The proportion of solvent systems used to obtain fraction 5, 7 and 15 were petroleum ether-EtOAc (95 : 5), (92 : 8) and (77 : 23) respectively. Fraction 5 gave betulinic acid (1, 20 mg) and fraction 7 gave lupeol (2, 10 mg) upon multiple pTLC using petroleum ether-EtOAc (95 : 5) and (90 : 10) respectively. pTLC of fraction 15 using petroleum ether-EtOAc (80 : 20) afforded betulinaldehyde (3, 15 mg).

Betulinic acid (1) ν<sub>max</sub>: 3060, 1630, 880 cm<sup>-1</sup>. EIMS m/z (rel. Int.): 456 [M⁺] (5), 441[M⁺ - CH₃] (10), 438 [M⁺ - H₂O] (20), 426 [M⁺ - (15-15)] (10), 415 [M⁺ - C₂H₅] (25), 208 (10), 206 (8), 163 (80), 135 (63), 107 (60), 105 (40), 79 (53), 41 (100). The ¹H NMR [δ<sub>H</sub>: 0.65, 0.75, 0.90, 0.96, 0.98 and 1.65], vinyl methyl [δ<sub>H</sub>: 1.67 (br d, J=0.5 Hz)], a secondary carbinol [δ<sub>H</sub>: 3.16 (dd, J=9.5, 6.0 Hz)] and [δ<sub>H</sub>: 2.95 (ddd, J=9.5, 6.0 Hz, 0.5 Hz )], an exomethylene group [δ<sub>H</sub>: 4.55 (1H,d, J=0.4 Hz )] and [δ<sub>H</sub>: 4.65 (1H,d, J=0.4 Hz)]. ¹³C NMR: 39.0 (C-1); 27.6 (C-2); 78.2 (C-3); 39.0 (C-4); 55.5 (C-5); 18.4 (C-6); 34.5 (C-7); 40.8 (C-8); 50.7 (C-9); 37.3 (C-10); 21.0 (C-11); 25.6 (C-12); 38.2 (C-13); 42.5 (C-14); 30.4 (C-15); 32.6 (C-16); 56.3 (C-17); 47.1 (C-18); 49.4 (C-19); 150.0 (C-20); 29.9 (C-21); 37.3 (C-22); 27.9 (C-23); 15.4 (C-24); 16.2 (C-25); 16.3 (C-26); 14.6 (C-27); 180.6 (C-28); 108.8 (C-29); 19.6 (C-30).

Lupeol (2), white crystals (MeOH), mp 210-212°C; [α]<sub>d</sub> +30.4 (C, 0.58 in CHCl₃); IR ν<sub>max</sub>: 3610, 3070, 3015, 1640, 1520, 1217, 1020, 887 cm<sup>-1</sup>; EIMS m/z (rel. Int.): 426 [M⁺] (2), 411 [M⁺ - CH₃] (3), 408 [M⁺ - H₂O] (3), 218 (5), 207 (6), 189 (58), 163 (80), 135 (57), 107 (68), 105 (55), 79 (54), 41 (100); ¹H NMR: δ<sub>H</sub>: 0.75, 0.78, 0.81, 0.92, 0.94, 1.02 (Me-28, Me-23, Me-24, Me-25, Me-26, Me-27), 1.67 (3H, br d, J=0.5 Hz, Me-30), 3.18 (1H, dd, J=9.6, 6.2 Hz, Hα-3), 4.56 (1H, d, J=0.4 Hz, Ha-29), 4.67 (1H, dq, J=0.4, 0.5 Hz, Hb-29); ¹³C NMR: δ<sub>C</sub>: 38.0 (C-1), 27.4 (C-2), 79.0 (C-3), 38.7 (C-4), 55.3 (C-5), 55.3 (C-5), 18.3 (C-5), 18.3 (C-6), 34.2 (C-7), 40.1 (C-8), 50.4 (C-9), 37.7 (C-10), 20.9 (C-11), 25.1 (C-12), 38.0 (C-13), 42.8 (C-14), 27.4 (C-15), 35.6 (C-16), 42.8 (C-17), 48.2 (C-18), 48.2 (C-18), 48.0 (C-19), 150.9 (C-20), 28.5 (C-21), 40.0 (C-22), 28.1 (C-23), 15.4 (C-24), 16.1 (C-25), 15.9 (C-26), 14.6 (C-27), 18.0 (C-28), 109.5 (C-29), 19.4 (C-30).

Betulinaldehyde (3), white crystals (MeOH), mp 188-190°C ν<sub>max</sub>: 3300, 2890, 1700, 1640, 885 cm<sup>-1</sup>, C₃₀H₄₄O₂ m/z 440 [M⁺] (10), 425 [M⁺ -15] (20), 422 [M⁺ -18] (55); 411 [M⁺ -CHO] (15), 407 [M⁺ -18-15] (20), 309 (10), 302 (15), 220 (15), 163 (80), 135 (63), 107 (60), 105 (40), 79 (53), 41 (100). ¹H NMR: δ<sub>H</sub>: 0.70, 0.80, 0.85, 0.90, 1.20 and 1.60, [δ<sub>H</sub>: 1.67 (br d, J=0.5 Hz)], [δ<sub>H</sub>: 3.17 (dd, J=9.5, 6.1 Hz)] and [δ<sub>H</sub>: 2.95 (ddd, J=9.5, 6.0 Hz, 0.5 Hz )], an exomethylene group [δ<sub>H</sub>: 4.55 (1H,d, J=0.4 Hz )] and [δ<sub>H</sub>: 4.65 (1H,d, J=0.4 Hz)]. ¹³C NMR: δ<sub>C</sub>: 39.1 (C-1); 27.6 (C-2); 79.0 (C-3); 39.1 (C-4); 55.4 (C-5); 18.3 (C-6); 34.4 (C-7); 40.7 (C-8); 50.6 (C-9); 37.7 (C-10); 20.9 (C-11); 25.5 (C-12); 38.1 (C-13); 42.4 (C-14); 30.5 (C-15); 32.5 (C-16); 56.2 (C-17), 47.0 (C-18); 49.3 (C-19); 150.0 (C-20); 29.8 (C-21); 37.2 (C-22); 27.9; (C-23); 15.4 (C-24); 16.2 (C-25); 16.3 (C-26); 14.6 (C-27); 180.0 (C-28); 108.8 (C-29); 19.6 (C-30).

**RESULTS AND DISCUSSION**

The ethyl acetate extract of the stem bark of *A. officinalis* afforded three triterpenoids (1-3). The isolated compounds were identified by spectroscopic analysis as well as by comparison of their spectral data with previously reported values.
Betulinic acid (1) was isolated as white crystal (MeOH). IR spectrum exhibited hydroxyl [νmax: 3610, 1020 cm⁻¹] and exomethylene [νmax: 3060, 1630, 880]. It mass spectrum displayed an [M⁺] peak at m/z 456 corresponding to C₃₀H₄₈O₃, together with fragments at m/z 441 [M⁺-15] and 438 [M⁺-18] and a base peak at m/z 43 [C₁₃H₁₇⁺].

The ¹H NMR spectrum of (1) revealed signals for five tertiary methyl. [δH: 0.65, 0.75, 0.90, 0.96, 0.98], a vinyl methyl [δH: 1.97 (br d, J=0.5 Hz)], a secondary carbinol [δH: 3.16 (dd, J=9.5 and 6.0 Hz)] and [δH: 2.95 (ddd, J=9.0, 6.0 and 0.5 Hz)] an exomethylene group [δH: 4.55 (1H, d, J=0.4 Hz)] and [δH: 4.65 (1H, d, J=0.4 Hz)]. These data indicated a pentacyclic triterpenoid of betulinic acid and comparison with published data³ confirmed the indentify of (1) as betulinic acid.

The ¹³C NMR spectrum of (3) showed six methyl group [δC: 27.9 (C-23), 15.4 (C-24), 16.2 (C-25), 16.3 (C-26), 14.6 (C-27), 19.6 (C-30)] and exomethylene group [δC: 150.0 (C-30), 108.8 (C-29)] and a secondary hydroxy-bearing carbon [δC: 79.0 (C-3)] and an carboxyl group at δC: 180.6 (C-28) in addition to ten methylene, five methine and five quaternary carbons. These data were identical to those reported betulinic acid.³

Lupeol (2) was isolated as white crystall from methanol and gave mp 210-212° [α]D + 30.4° (C, 0.58 in CHCl₃). Its IR spectrum exhibited hydroxyl [νmax: 3610, 1020 cm⁻¹] and exomethylene [νmax: 3070, 1640, 887 cm⁻¹] absorption. The mass spectrum displayed a molecular ion [M⁺] peak at m/z 426 corresponding to C₃₀H₅₀O together with fragments at m/z 411 [M⁺-15] and 408 [M⁺-18] which were due to the loss of methyl group and a molecule of water from the molecular ion peak. The mass spectrum also showed a base peak at m/z 41 [C₁₃H₁₇⁺] arising from the loss of the side chain of lupeol. The ¹H NMR spectrum exhibited six tertiary methyl singlets at [δH: 0.75, 0.77, 0.80, 0.92, 0.94 and 1.02], a methine group at [δH: 1.66 (br d, J=0.5 Hz)], a secondary carbinol group at [δH: 3.20 (dd, J=9.6 and 6.2 Hz)] and an exomethylene group at [δH: 4.58 (1H, d, J=0.4 Hz) and [δH: 4.65 (1H, dq, J=0.4 and 0.5 Hz)] typical of pentacyclic triterpenoid of the lupeol (1).

The structural assignment of (2) was further substantiated by its ¹³C NMR spectrum which showed seven methyl groups at [δC: 28.0 (C-23), 19.3 (C-30), 18.0 (C-28), 16.1 (C-25), 15.9 (C-26), 15.4
(C-24), 14.5 (C-27)], an exomethylene group at $[\delta_c: 150.8 \text{ (C-20)}, 109.3 \text{ (C-29)}]$ and a secondary hydroxyl bearing carbon at $[\delta_c: 78.9 \text{ (C-3)}]$, in addition to ten methylene, five methine and five quaternary carbons. The shielding of C-23 methyl of (2) could be due to the influence of the adjacent C-3 hydroxyl group.$^{3,6}$ These data were in close agreement with those reported for lupeol (2)$^{3,6}$ and further confirmed the identity of (2) as lupeol.

Betulinaldehyde (3) was isolated as crystals (MeOH), mp188-190$^\circ$. It IR spectrum displayed absorption at $\nu_{\text{max}}: 3300, 2890, 1700, 1640, 885 \text{ cm}^{-1}$. It’s exhibited a [M$^+$] peak at m/z 440 corresponding to C$_{30}$H$_{48}$O$_2$, together with fragments at m/z 425, [M$^+$ - 15] and 410 [M$^+$ -18] and a base peak at m/z 41 [C$_{3}$H$_{3}^-$] corresponding to a lupeol type triterpinoid.

The $^1$H NMR spectrum of (3) revealed signals for five tertiary methyl. [$\delta_H: 0.70, 0.80, 0.85, 0.90, 1.20$ and 1.60] a vinyl methyl [$\delta_H: 1.67$ (br d, $J=0.5 \text{ Hz}$)] a secondary carbinol [$\delta_H: 3.17$ (dd, $J=9.5$ and 6.1 Hz)] and [$\delta_H: 2.95$ (ddd, $J=9.5, 6.0$ and 0.5 Hz)] an exomethylene group [$\delta_H: 4.55$ (1H, d, $J=0.4 \text{ Hz}$)] and [$\delta_H: 4.65$ (1H, d, $J=0.4 \text{ Hz}$)]. These data indicated a pentacyclic triterpinoid of lupeol type with an aldehyde group and comparation with published data$^6$ confirmed the identity of (3) as betulinaldehyde.

The $^{13}$C NMR spectrum of (3) showed six methyl groups [$\delta_c: 27.9 \text{ (C-23)}, 15.4 \text{ (C-24)}, 16.2 \text{ (C-25)}, 16.3 \text{ (C-26)}, 14.6 \text{ (C-27)}, 19.6 \text{ (C-30)}$] and an
exomethylene group \( [\delta_C: 150.0 \text{ (C-20), } 108.8 \text{ (C-29)}] \) and a secondary hydroxyl bearing carbon \( [\delta_C: 79.0 \text{ (C-3)}] \), and an aldehyde group at \( [\delta_C: 180.0 \text{ (C-28)}] \), in addition to ten methylene, five methine and five quaternary carbons. These data were identical to those of betulinaldehyde.\(^6\) This is the first report of the isolation of these triterpinoids from *Avicennia Officinalis*. Further analysis may result in the isolation of more biologically active compounds.

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**REFERENCES**